

- 1 -

COMPOSITIONS FOR WASHING AND CONDITIONING HAIRTechnical Field

5

The invention is concerned with rinse-off hair-conditioning compositions which are applied to the hair or body and then substantially rinsed away. It is particularly concerned with hair shampoo compositions and shower gels, which both
10 clean the hair and provide conditioning benefit to the hair. More specifically, it is concerned with hair-washing compositions which have the effect of providing low friction and ease of combing to the hair when dry.

15 Background and Prior Art

Compositions which provide a combination of cleansing and conditioning to the hair are known in the art. Such shampoo or shower-gel compositions typically comprise one or more
20 surfactants for shampooing or cleansing purposes and one or more conditioning agents. Typically, these conditioning agents are water-insoluble oily materials, cationic polymers or cationic surfactants. One purpose of the conditioning agent is to make the hair easier to comb when wet and more
25 manageable when dry, e.g. less static and fly-away. Another important role, especially for water-insoluble oily conditioning agents, is to provide low friction and ease of combing for dry hair.

30 It is known to incorporate cationic polymers in hair-washing compositions. For instance, US 6,444,628 discloses an

- 2 -

aqueous shampoo comprising, in addition to water, an anionic cleansing surfactant, a cationic polymer and a monoalkyl quaternary ammonium compound.

5 Such cationic polymers are often used in combination with water-insoluble conditioning oils in order to improve the deposition of the conditioning oils onto the hair. US Patent 3,753,916 discloses the use of cationic polymers as deposition aids.

10

It has now been found that by using a specific combination of cationic polymers in a hair-washing compositions which comprise small droplets of a water-insoluble oily conditioning agent, problems encountered with prior art washing and conditioning compositions can be overcome. In particular, improved low friction and ease of combing can be obtained for the hair after drying, with a reduction in the heavy, greasy feel that many consumers experience when high charge density cationic polymers and oily conditioning agents are combined in shampoos.

15
20

Summary of the Invention

In a first aspect, the invention provides a hair-washing composition comprising

25

a) from 1 to 50% by weight of a cleansing surfactant,

30

b) from 0.01 to 0.5% by weight of a first cationic polymer having a mean charge density at pH7 from 0.2 to 1.0 meq per gram,

- 3 -

c) from 0.01 to 0.4% by weight of a second cationic polymer having a mean charge density at pH7 from 1.3 to 3.0 meq per gram,

5 d) more than 40% by weight of water, and

e) from 0.1 to 10% by weight of discrete, dispersed droplets of a water-insoluble conditioning oil with a mean diameter ($D_{3,2}$) of 4 micrometres or less,

10

characterised in that both the first cationic polymer and the second cationic polymer consist essentially of the same monomeric units.

15 Detailed Description of the Invention

By water-insoluble, it is meant that the material so described has a solubility in water at 25°C of 0.1% by weight or less.

20

All viscosities mentioned are kinematic viscosities unless otherwise specified, and are to be measured at 25°C using calibrated capillary glass viscometers under gravity flow conditions.

25

All molecular weights referred to are weight average (M_w) molecular weights unless otherwise specified.

The compositions provided by the invention are aqueous compositions, used by massaging them into the hair followed
30 by rinsing with clean water prior to drying the hair.

- 4 -

Optionally, a separate conditioning formulation may be applied after rinsing and before drying, but this may not be necessary as the compositions of the invention are intended to provide both cleansing and conditioning to the hair.

5

The compositions provided by the invention comprise more than 40% by weight of water, preferably more than 50%, more preferably more than 65%.

10 Cleansing Surfactant

Hair-washing compositions according to the invention comprise one or more cleansing surfactants from the group which is cosmetically acceptable and suitable for topical
15 application to the hair.

Suitable cleansing surfactants, which may be used singularly or in combination, are selected from anionic, nonionic, amphoteric and zwitterionic surfactants, and mixtures
20 thereof.

Anionic Cleansing Surfactant

Shampoo compositions according to the invention will
25 typically comprise one or more anionic cleansing surfactants which are cosmetically acceptable and suitable for topical application to the hair.

Examples of suitable anionic cleansing surfactants are the
30 alkyl sulphates, alkyl ether sulphates, alkaryl sulphonates, alkanoyl isethionates, alkyl succinates, alkyl

Co-surfactant

The composition can include co-surfactants, to help impart aesthetic, physical or cleansing properties to the
5 composition.

A preferred example is an amphoteric or zwitterionic surfactant, which can be included in an amount ranging from 0 to about 8, preferably from 1 to 4 percent by weight of the
10 composition.

Examples of amphoteric and zwitterionic surfactants include alkyl amine oxides, alkyl betaines, alkyl amidopropyl betaines, alkyl sulphobetaines (sultaines), alkyl glycinate,
15 alkyl carboxyglycinates, alkyl amphopropionates, alkylamphoglycinates, alkyl amidopropyl hydroxysultaines, acyl taurates and acyl glutamates, wherein the alkyl and acyl groups have from 8 to 19 carbon atoms. Typical amphoteric and zwitterionic surfactants for use in shampoos of the
20 invention include lauryl amine oxide, cocodimethyl sulphopropyl betaine and preferably lauryl betaine, cocamidopropyl betaine and sodium cocamphopropionate.

Another preferred example is a nonionic surfactant, which can
25 be included in an amount ranging from 0 to 8, preferably from 2 to 5 percent by weight of the composition.

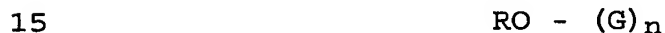
For example, representative nonionic surfactants that can be included in shampoo compositions of the invention include
30 condensation products of aliphatic (C₈ - C₁₈) primary or secondary linear or branched chain alcohols or phenols with

- 7 -

alkylene oxides, usually ethylene oxide and generally having from 6 to 30 ethylene oxide groups.

Other representative nonionic surfactants include mono- or
5 di-alkyl alkanolamides. Examples include coco mono- or di-ethanolamide and coco mono-isopropanolamide.

Further nonionic surfactants which can be included in shampoo compositions of the invention are the alkyl polyglycosides
10 (APGs). Typically, the APG is one which comprises an alkyl group connected (optionally via a bridging group) to a block of one or more glycosyl groups. Preferred APGs are defined by the following formula:



wherein R is a branched or straight chain alkyl group which may be saturated or unsaturated and G is a saccharide group.

20 R may represent a mean alkyl chain length of from about C₅ to about C₂₀. Preferably R represents a mean alkyl chain length of from about C₈ to about C₁₂. Most preferably the value of R lies between about 9.5 and about 10.5. G may be selected from C₅ or C₆ monosaccharide residues, and is preferably a
25 glucoside. G may be selected from the group comprising glucose, xylose, lactose, fructose, mannose and derivatives thereof. Preferably G is glucose.

- 8 -

The degree of polymerisation, n , may have a value of from about 1 to about 10 or more. Preferably, the value of n lies from about 1.1 to about 2. Most preferably the value of n lies from about 1.3 to about 1.5.

5

Suitable alkyl polyglycosides for use in the invention are commercially available and include for example those materials identified as: Oramix NS10 ex Seppic; Plantaren 1200 and Plantaren 2000 ex Henkel.

10

Other sugar-derived nonionic surfactants which can be included in compositions of the invention include the C_{10} - C_{18} N-alkyl (C_1 - C_6) polyhydroxy fatty acid amides, such as the C_{12} - C_{18} N-methyl glucamides, as described for example in WO 92 06154 and US 5 194 639, and the N-alkoxy polyhydroxy fatty acid amides, such as C_{10} - C_{18} N-(3-methoxypropyl) glucamide.

15

The composition according to the invention can also optionally include one or more cationic co-surfactants included in an amount ranging from 0.01 to 10, more preferably from 0.05 to 5, most preferably from 0.05 to 2 percent by weight of the composition.

20

The total amount of cleansing surfactant (including any co-surfactant, and/or any emulsifier) in compositions of the invention is generally from 1 to 50, preferably from 2 to 40, more preferably from 10 to 25 percent by weight of composition.

25

30

- 9 -

A preferred blend of cleansing surfactants is a combination of ammonium lauryl ether sulphate, ammonium lauryl sulphate, PEG 5 cocamide and cocamide MEA (CTFA designations).

5 Cationic Polymer

Compositions according to the invention comprise from 0.01 to 0.5% by weight of a first cationic polymer having a mean charge density at pH7 from 0.2 to 1.0 meq per gram, and
10 from 0.01 to 0.4% by weight of a second cationic polymer having a mean charge density at pH7 from 1.3 to 3.0 meq per gram.

It is essential for the invention that both the first and
15 second cationic polymer consist essentially of the same monomers and cationic substituents, by which it is meant that the polymers comprise 95% or more by weight of the same monomers. When the polymers are cationically substituted homopolymers or copolymers, then the cationic substituent
20 must be essentially the same for each polymer, but the degree of substitution will be different for the first and the second homopolymer in order to give the required charge densities for the two cationic polymers.

25 The cationic polymers may be homopolymers which are cationically substituted or may be formed from two or more types of monomers. The molecular weight of each polymer will generally be between 100 000 and 2 000 000 Dalton. The polymers will have cationic nitrogen containing groups such
30 as quaternary ammonium or protonated amino groups, or a mixture thereof. If the molecular weight of the polymer is

- 10 -

too low, then the conditioning effect is poor. If too high, then there may be problems of high extensional viscosity leading to stringiness of the composition when it is poured.

- 5 The cationic nitrogen-containing group will generally be present as a substituent on a fraction of the total monomer units of the cationic polymer. Thus when the polymer is not a homopolymer it can contain spacer non-cationic monomer units. Such polymers are described in the CTFA Cosmetic
- 10 Ingredient Directory, 3rd edition. The ratio of the cationic to non-cationic monomer units is selected to give polymers having a cationic charge density in the required range for each of the first and second cationic polymers.
- 15 Suitable cationic polymers include, for example, copolymers of vinyl monomers having cationic amine or quaternary ammonium functionalities with water soluble spacer monomers such as (meth)acrylamide, alkyl and dialkyl
- 20 (meth)acrylamides, alkyl (meth)acrylate, vinyl caprolactone and vinyl pyrrolidine. The alkyl and dialkyl substituted monomers preferably have C1-C7 alkyl groups, more preferably C1-3 alkyl groups. Other suitable spacers include vinyl esters, vinyl alcohol, maleic anhydride, propylene glycol and ethylene glycol.
- 25
- The cationic amines can be primary, secondary or tertiary amines, depending upon the particular species and the pH of the composition. In general secondary and tertiary amines, especially tertiary, are preferred.

- 11 -

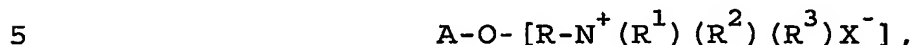
Amine substituted vinyl monomers and amines can be polymerized in the amine form and then converted to ammonium by quaternization.

- 5 The cationic conditioning polymers can comprise mixtures of monomer units derived from amine- and/or quaternary ammonium-substituted monomer and/or compatible spacer monomers.
- 10 Suitable cationic conditioning polymers include, for example:
- cationic diallyl quaternary ammonium-containing polymers including, for example, dimethyldiallylammonium chloride
15 homopolymer and copolymers of acrylamide and dimethyldiallylammonium chloride, referred to in the industry (CTFA) as Polyquaternium 6 and Polyquaternium 7, respectively;
 - 20 - mineral acid salts of amino-alkyl esters of homo- and copolymers of unsaturated carboxylic acids having from 3 to 5 carbon atoms, (as described in U.S. Patent 4,009,256);
 - 25 - cationic polyacrylamides (as described in WO95/22311).

Other cationic conditioning polymers that can be used include cationic polysaccharide polymers, such as cationic cellulose derivatives, cationic starch derivatives, and cationic guar
30 gum derivatives.

- 12 -

Cationic polysaccharide polymers suitable for use in compositions of the invention include monomers of the formula:



wherein: A is an anhydroglucose residual group, such as a starch or cellulose anhydroglucose residual. R is an alkylene, oxyalkylene, polyoxyalkylene, or hydroxyalkylene
10 group, or combination thereof. R^1 , R^2 and R^3 independently represent alkyl, aryl, alkylaryl, arylalkyl, alkoxyalkyl, or alkoxyaryl groups, each group containing up to about 18 carbon atoms. The total number of carbon atoms for each cationic moiety (i.e., the sum of carbon atoms in R^1 , R^2 and
15 R^3) is preferably about 20 or less, and X is an anionic counterion.

Another type of cationic cellulose includes the polymeric quaternary ammonium salts of hydroxyethyl cellulose reacted
20 with lauryl dimethyl ammonium-substituted epoxide, referred to in the industry (CTFA) as Polyquaternium 24. These materials are available from Amerchol Corp. (Edison, NJ, USA) for instance under the tradename Polymer LM-200.

25 Other suitable cationic polysaccharide polymers include quaternary nitrogen-containing cellulose ethers (e.g. as described in U.S. Patent 3,962,418), and copolymers of etherified cellulose and starch (e.g. as described in U.S. Patent 3,958,581).

- 13 -

A particularly suitable type of cationic polysaccharide polymer that can be used is a cationic guar gum derivative, such as guar hydroxypropyltrimonium chloride (commercially available from Rhone-Poulenc in their JAGUAR trademark series).

The first cationic polymer will generally be present in compositions of the invention at levels of from 0.01 to 0.5, preferably from 0.05 to 0.4, more preferably from 0.1 to 0.3 percent by weight of the composition.

The second cationic polymer will generally be present in compositions of the invention at levels of from 0.01 to 0.4, preferably from 0.05 to 0.35, more preferably from 0.1 to 0.3 percent by weight of the composition.

The cationic charge density of the polymer is suitably determined via the Kjeldahl method as described in the US Pharmacopoeia under chemical tests for nitrogen determination.

It is preferred if the first and cationic polymers are both cationically substituted guar gums, both cationically substituted hydroxyethylcelluloses or both cationic polyacrylamides, because of the relative ease of manufacture and wide availability of these polymers.

It is particularly preferred if both the first and second cationic polymers are guar hydroxypropyltrimethylammonium chlorides.

- 14 -

The first cationic polymer has a mean charge density from 0.2 to 1.0 meq/gm, preferably from 0.3 to 0.9 meq/gm.

5 The second cationic polymer has a mean charge density from 1.3 to 3 meq/gm, preferably from 1.4 to 2.5, more preferably from 1.5 to 1.8.

10 If the charge density of the second cationic polymer is higher than the specified range, it is found that the hair may feel heavy or greasy. This is thought to be due to excessive deposition of the second polymer. If the charge density of the first polymer is lower than the specified range, it is found that the wet conditioning of the composition may be inadequate.

15

Water-insoluble Conditioning Oil

20 Compositions according to the invention comprise from 0.1 to 10% by weight of a water-insoluble conditioning oil. This may be a non-silicone hydrophobic oil but is more preferably a silicone conditioning agent. Preferably the conditioning agent is non-volatile, meaning that it has a vapour pressure of less than 1000 Pa at 25°C. The conditioning oil is present in the composition as discrete emulsion droplets.

25

The total amount of water-insoluble conditioning oil in compositions of the invention is preferably from 0.2% to 5%, more preferably from 0.5% to 3% by weight of the total composition.

30

- 15 -

Emulsified hydrophobic conditioning oils for use in the shampoo or shower gel compositions of the invention suitably have Sauter mean droplet diameter ($D_{3,2}$) in the composition of 4 micrometres or less, preferably 2 micrometres or less, more preferably 1 micrometre or less.

A suitable method for measuring the Sauter $D_{3,2}$ mean diameter is by laser light scattering using an instrument such as a Malvern Mastersizer.

10

Silicone Conditioning Oil

It is preferred if the water-insoluble conditioning oil of compositions of the invention is emulsified droplets comprising, preferably consisting essentially of, a silicone conditioning oil.

Suitable silicones include polydiorganosiloxanes, in particular polydimethylsiloxanes which have the CTFA designation dimethicone. Also suitable for use compositions of the invention (particularly shampoos and conditioners) are polydimethyl siloxanes having hydroxyl end groups, which have the CTFA designation dimethiconol. Also suitable for use in compositions of the invention are silicone gums having a slight degree of cross-linking, as are described for example in WO 96/31188.

The viscosity of the silicone itself (not the emulsion or the final hair conditioning composition) is typically from 350 to 200,000,000 $\text{mm}^2 \text{sec}^{-1}$ at 25°C. Preferably the viscosity is at

30

- 16 -

least $5,000 \text{ mm}^2 \text{ sec}^{-1}$ at 25°C , more preferably at least $10,000 \text{ mm}^2 \text{ sec}^{-1}$. Preferably the viscosity does not exceed $20,000,000 \text{ mm}^2 \text{ sec}^{-1}$, more preferably $10,000,000 \text{ mm}^2 \text{ sec}^{-1}$, most preferably $5,000,000 \text{ mm}^2 \text{ sec}^{-1}$.

5

Viscosities are generally provided by suppliers of silicones, either as measured or as deduced from their molecular weight.

It is preferred if the silicone oil also comprises a
10 functionalised silicone. Suitable functionalised silicones include, amino-, carboxy-, betaine-, quaternary ammonium-, carbohydrate-, hydroxy- and alkoxy-substituted silicones. Preferably, the functionalised silicone contains multiple substitutions.

15

For the avoidance of doubt, as regards hydroxyl-substituted silicones, a polydimethylsiloxane merely having hydroxyl end groups (which have the CTFA designation dimethiconol) is not considered a functionalised silicone within the definition
20 of the present invention. However, a polydimethylsiloxane having hydroxyl substitutions along the polymer chain is considered a functionalised silicone.

A preferred class of functionalised silicone for inclusion
25 in compositions of the invention is amino functional silicone. By "amino functional silicone" is meant a silicone containing at least one primary, secondary or tertiary amine group, or a quaternary ammonium group. Examples of suitable amino functional silicones include:
30 polysiloxanes having the CTFA designation "amodimethicone",

- 17 -

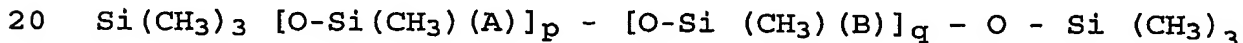
Specific examples of amino functional silicones suitable for use in the invention are the aminosilicone oils DC2-8220, DC2-8166, DC2-8466, and DC2-8950-114 (all ex Dow Corning), and GE 1149-75, (ex General Electric Silicones).

5 Suitable quaternary silicone polymers are described in EP-A-0 530 974. A preferred quaternary silicone polymer is K3474, ex Goldschmidt.

Another preferred functional silicone for use as a component
10 in the hydrophobic conditioning oil is an alkoxy-substituted silicone. Such molecules are known as silicone copolyols and have one or more polyethyleneoxide or polypropyleneoxide groups bonded to the silicone polymer backbone, optionally through an alkyl linking group.

15

An example of a type of silicone copolyol useful in compositions of the invention has a molecular structure according to the formula depicted below:



In this formula, A is an alkylene chain with from 1 to 22 carbon atoms, preferably 4 to 18, more preferably 10 to 16.

B is a group with the structure: $-(\text{R})-(\text{EO})_r(\text{PO})_s-\text{OH}$ wherein

25 R is a linking group, preferably an alkylene group with 1 to 3 carbon atoms. Preferably R is $-(\text{CH}_2)_2-$. The mean values of r and s are 5 or more, preferably 10 or more, more preferably 15 or more. It is preferred if the mean values of r and s are 100 or less. In the formula, the value of p
30 is suitably 10 or more, preferably 20 or more, more

- 18 -

preferably 50 or more and most preferably 100 or more. The value of q is suitably from 1 to 20 wherein the ratio p/q is preferably 10 or more, more preferably 20 or more. The value of $p + q$ is a number from 11 to 500, preferably from
5 50 to 300.

Suitable silicone copolyols have an HLB of 10 or less, preferably 7 or less, more preferably 4 or less. A suitable silicone copolyol material is DC5200, known as Lauryl
10 PEG/PPG - 18/18 methicone (INCI name), available from Dow Corning.

Hydrophile/Lipophile balance or HLB is a well known parameter used by those skilled in the art to characterise
15 surface active molecules and emulsifiers. Suitable methods for the experimental determination of HLB are in Griffin W.C, Journal of the Society of Cosmetic Chemists, volume 1 page 311 (1949). The commercially available silicone copolyols are supplied along with a value of their HLB by
20 Dow Corning.

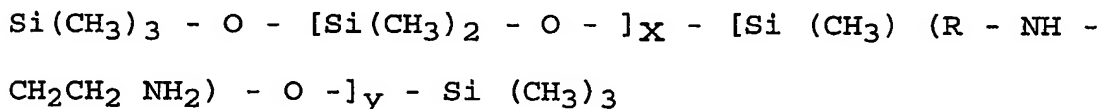
It is preferred to use a combination of amino and non-functional silicones. In particular, when the water-insoluble oil is a silicone oil blend, it is preferred if
25 the silicone oil blend comprises

(i) from 50 to 95% by weight of the total weight of silicone oil of a polydimethylsiloxane gum having a molecular weight of 200,000 unified mass units or more and

- 19 -

(ii) from 5 to 50% by weight of the total weight of silicone oil a second silicone which is an amino-functionalised polydimethylsiloxane having a molecular weight of less than 200,000 unified mass units.

Suitable amino-functionalised silicones for such a blend are described in EP 455,185 (Helene Curtis) and include trimethylsilylamodimethicone as depicted below, and are sufficiently water-insoluble so as to be useful in compositions of the invention:



wherein $x + y$ is a number from about 50 to about 500, and the mole percent amine functionality is from 0.3% to 8%, and wherein R is an alkylene group having from 2 to 5 carbon atoms. Preferably, the number $x + y$ is from 100 to 300, and the mole percent amine functionality is from 0.3% to 8%.

As expressed here, the weight percent amine functionality is measured by titrating a sample of the amino-functionalised silicone against alcoholic hydrochloric acid to the bromocresol green end point. The mole percent amine is calculated using a molecular weight of 45 (corresponding to $\text{CH}_3 - \text{CH}_2 - \text{NH}_2$).

- 20 -

Suitably, the mole percent amine functionality measured and calculated in this way is from 0.3% to 8%, preferably from 0.5% to 4%.

- 5 An example of a commercially available amino-functionalised silicone useful in the silicone component of the composition of the invention is DC-8220 available from Dow Corning, which has a viscosity of $150 \text{ mm}^2 \text{ s}^{-1}$ at 25°C and a mole percent amine functionality of 2.0%.

10

Polydimethylsiloxane silicone gums are also a component of the preferred silicone oil blend described above.

15

The polydimethylsiloxane gum is present at a level of at least 50 wt%, preferably at least 60 wt% based on the total weight of the silicone component.

20

The polydimethylsiloxane gum suitably has a viscosity of at least 500,000 mm^2/sec , preferably at least 600,000 mm^2/sec , more preferably at least 1,000,000 mm^2/sec at 25°C .

25

Suitably, the first silicone has a molecular weight of at least 200,000 Dalton preferably at least 400,000 Dalton, more preferably at least 500,000 Dalton.

Suitable silicone gums include SE30, SE54 and SE76 (ex General Electric Silicones).

30

The silicones may be added to the composition as a fluid and subsequently emulsified, but preferably are added as pre-

- 21 -

formed emulsions for ease of processing. Preferably, the pre-formed silicone emulsions additionally comprise a suitable emulsifier such as dodecylbenzenesulphonic acid, or are emulsified using the surface active block copolymer as the emulsifier. A preferred form of silicone oil emulsion is one which has been preferred by mechanical emulsification using a high shear mixer.

Non-silicone Conditioning Oil

10

Compositions according to the present invention may comprise a dispersed, non-volatile, water-insoluble oily non-silicone conditioning agent as the water-insoluble conditioning oil.

15 Suitable oily or fatty materials are selected from hydrocarbon oils, fatty esters and mixtures thereof.

Straight chain hydrocarbon oils will preferably contain from about 12 to about 30 carbon atoms. Also suitable are polymeric hydrocarbons of alkenyl monomers, such as C2-C6 alkenyl monomers.

Specific examples of suitable hydrocarbon oils include paraffin oil, mineral oil, saturated and unsaturated dodecane, saturated and unsaturated tridecane, saturated and unsaturated tetradecane, saturated and unsaturated pentadecane, saturated and unsaturated hexadecane, and mixtures thereof. Branched-chain isomers of these compounds, as well as of higher chain length hydrocarbons, can also be used. Another suitable material is polyisobutylene.

- 22 -

- Suitable fatty esters are characterised by having at least 10 carbon atoms, and include esters with hydrocarbyl chains derived from fatty acids or alcohols, Monocarboxylic acid esters include esters of alcohols and/or acids of the
- 5 formula $R'COOR$ in which R' and R independently denote alkyl or alkenyl radicals and the sum of carbon atoms in R' and R is at least 10, preferably at least 20. Di- and trialkyl and alkenyl esters of carboxylic acids can also be used.
- 10 Particularly preferred fatty esters are mono-, di- and triglycerides, more specifically the mono-, di-, and tri-esters of glycerol and long chain carboxylic acids such as C1-C22 carboxylic acids. Preferred materials include cocoa butter, palm stearin, sunflower oil, soyabean oil and
- 15 coconut oil.

- Preferably, the viscosity of the non-silicone conditioning oil itself is from 350 to 10,000,000 $\text{mm}^2\text{sec}^{-1}$ at 25°C. More preferably the viscosity is at least 5,000 $\text{mm}^2\text{sec}^{-1}$ at 25 °C,
- 20 most preferably at least 10,000 $\text{mm}^2\text{sec}^{-1}$. Preferably the viscosity does not exceed 500,000 $\text{mm}^2\text{sec}^{-1}$.

Other Ingredients

- 25 Compositions according to the invention may contain other ingredients suitable for use in hair cleansing and conditioning compositions. Such ingredients include but are not limited to: fragrance, suspending agents, amino acids

- 23 -

and protein derivatives, viscosity modifiers and preservatives.

The invention will now be further illustrated by reference
5 to the following, non-limiting examples.

Examples

Compositions were prepared according to the formulations
10 detailed in tables 1 and 2. All ingredients are expressed
in the tables by weight percent of the total formulation,
and as level of active ingredient.

Table 1

Ingredient	Weight Percent Active
Anionic Surfactant	10
Amphoteric Surfactant (CAPB)	3
Cationic Guar (See table 2)	0.2
Emulsified silicone	1.5
Minors and Water	to 100

15 The anionic surfactant is an Ether Sulphate (1EO), CAPB is
coco amidopropyl betaine. The cationic guar is either
Jaguar C17 or Jaguar C13S or a combination of the two as
detailed below in table 2. The emulsified silicone is
20 droplets of a aminosilicone (mol wt 32,000 and mole % amine
content of about 1.75%) and polydimethylsiloxane gum (mol wt
700,000 u) with a mean $D_{3,2}$ droplet diameter of 1 micron.

Table 2

Example	JaguarC13S	JaguarC17	Friction	Deposition
A	0.2	0	25643	108
1	0.175	0.025	22827	301
2	0.15	0.05	18896	515
3	0.125	0.075	16284	583
4	0.1	0.1	18098	615
B	0	0.2	15576	659

0.25g/5 cm switches of hair which had been cleaned with a
5 solution of 14% SLES 2EO and 2% cocoamidopropyl betaine in
water followed by extensive rinsing, were used as the basis
for the following analysis. The test shampoo was diluted to
1 in 10 by weight with distilled water and stirred throughout
10 of a petri dish. 1.5 mls of diluted shampoo was placed along
the length of the switches which were then agitated in the
dish for 30 seconds, followed by a rinse for 30 seconds under
tap water (12° French hard) at 40°C, with a flow rate set at
3-4 litres per minute. The washing process using the test
15 shampoo solution was repeated followed again by rinsing. The
switches were then allowed to dry naturally at 25° C and a
relative humidity of 45 to 60%.

The amount of silicone deposited on the hair samples was
20 measured using X-ray fluorescence spectrometry (measured in
parts per million (ppm) of elemental silicon.

- 25 -

The friction was measured using a laboratory technique with a commercially available texture analyser TA XT2i ex Stable Microsystems.

5 Compositions according to Examples B and 3 were each compared in a home-use test against a benchmark formulation with only 0.2% Jaguar C13S corresponding to formula A. Each formulation was scored by the users for a set of performance attributes.

10

Both the examples B and 3 were significantly preferred over the benchmark in the home-use test for the attribute "smooth feel" of the hair (90% and 95% significance respectively). However, Example B gave a significantly (90%) poorer score
15 than the benchmark for "Greasy Scalp". By contrast, example 3 was preferred over the benchmark (directionally but not significantly) on greasy scalp.

The home-use test results, and the friction and deposition
20 test results, demonstrate that compositions according to the invention are able to achieve conditioning performance equivalent to compositions with a high charge density cationic polymer alone, yet without the negative of greasy scalp feel.

- 26 -

CLAIMS

1. A hair-washing composition comprising

- 5 a) from 1 to 50% by weight of a cleansing surfactant,
- b) from 0.01 to 0.5% by weight of a first cationic
polymer having a mean charge density at pH7 from 0.2
to 1.0 meq per gram,
- 10 c) from 0.01 to 0.4% by weight of a second cationic
polymer having a mean charge density at pH7 from 1.3
to 3.0 meq per gram,
- 15 d) more than 40% by weight of water, and
- e) from 0.1 to 10% by weight of discrete, dispersed
droplets of a water-insoluble conditioning oil with
a mean diameter ($D_{3,2}$) of 4 micrometres or less,

20

characterised in that both the first cationic polymer and
the second cationic polymer consist essentially of the same
monomeric units and cationic substituents.

- 25 2. A composition according to claim 1 wherein the first and
second cationic polymers are both cationically substituted
guar gums.

- 27 -

3. A composition according to claim 2 wherein the first and second cationic polymers are both guar hydroxypropyltrimethylammonium chlorides.

5 4. A composition according to claim 1 wherein the first and second cationic polymers are both cationically substituted hydroxyethyl celluloses.

10 5. A composition according to claim 1 wherein the first and second cationic polymers are both cationic polyacrylamides.

15 6. A composition according to any preceding claim wherein the cleansing surfactant is selected from the group consisting of anionic surfactants, amphoteric surfactants, zwitterionic surfactants, nonionic surfactants and mixtures thereof.

20 7. A composition according to any preceding claim wherein the water-insoluble conditioning oil comprises a silicone oil.

8. A composition according to claim 7 wherein the silicone oil comprises

25 (i) from 50 to 95% by weight of the total weight of silicone oil of a polydimethylsiloxane gum having a molecular weight of 200,000 unified mass units or more and

30 (ii) from 5 to 50% by weight of the total weight of silicone oil of a second silicone which is an

- 28 -

amino-functionalised polydimethylsiloxane having a molecular weight of less than 200,000 unified mass units.

- 5 9. A composition according to claim 8, in which the polydimethylsiloxane gum has a viscosity of 500,000 mm²/sec or more at 25°C
- 10 10. A composition according to claim 8 or claim 8, in which the amino-functional polydimethylsiloxane has a viscosity of less than 500,000 mm²/sec at 25°C.
- 15 11. A composition according to any one of claims 8 to 10, in which the amino-functionalised silicone has a mole percent amino functionality from 0.3 to 8.
- 20 12. A composition according to any of claims 8 to 11, in which the weight ratio of the polysdimethylsiloxane gum to the amino-functional polydimethylsiloxane is from 15:1 to 1:1.
- 25 13. A composition according to any preceding claim, in which the water-insoluble conditioning oil is in the form of a mechanically generated emulsion.
14. A method of washing and conditioning hair by massaging a composition according to any preceding claim into the hair followed by rinsing prior to drying the hair.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 03/12066

A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 A61K7/06

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 7 A61Q A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 529 883 A (UNILEVER PLC ; UNILEVER NV (NL)) 3 March 1993 (1993-03-03) page 2, line 24 - page 4, line 9; claims 1-10	1-4,6,7, 9-11,13, 14
A	WO 00/66081 A (PROCTER & GAMBLE) 9 November 2000 (2000-11-09) page 21, line 19 - page 24, line 21; claims 1-5; examples 1-15	1-4,6,7
A	US 6 040 282 A (FRITSCHI LYNNE ANNE ET AL) 21 March 2000 (2000-03-21) claims 1-20	1-4,6,7
A	US 5 543 074 A (MURRAY ANDREW M ET AL) 6 August 1996 (1996-08-06) column 1, line 22 - line 47; claims 1-6	1,5-7
	-/-	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the international search

18 February 2004

Date of mailing of the international search report

26/02/2004

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Giese, H-H

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 03/12066

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	<p>US 6 090 773 A (ALLAN WILLIAM D ET AL) 18 July 2000 (2000-07-18) column 3, line 50 - line 53; claim 1</p>	1-6

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 03/12066

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 0529883	A	03-03-1993	AT 147964 T	15-02-1997
			AU 651236 B2	14-07-1994
			AU 2104092 A	18-02-1993
			BR 9203124 A	30-03-1993
			CA 2075767 A1	17-02-1993
			DE 69216953 D1	06-03-1997
			DE 69216953 T2	12-06-1997
			EP 0529883 A1	03-03-1993
			ES 2098453 T3	01-05-1997
			GR 3023138 T3	30-07-1997
			IN 176963 A1	12-10-1996
			JP 5194142 A	03-08-1993
			JP 7080740 B	30-08-1995
			KR 9615951 B1	25-11-1996
			ZA 9206141 A	14-02-1994
WO 0066081	A	09-11-2000	AU 759514 B2	17-04-2003
			AU 4689100 A	17-11-2000
			CA 2370939 A1	09-11-2000
			CN 1353601 T	12-06-2002
			EP 1175202 A1	30-01-2002
			JP 2002543105 T	17-12-2002
			WO 0066081 A1	09-11-2000
			US 6451300 B1	17-09-2002
US 6040282	A	21-03-2000	WO 9938476 A1	05-08-1999
			AU 735002 B2	28-06-2001
			AU 2578399 A	16-08-1999
			CA 2317703 A1	05-08-1999
			CN 1289245 T	28-03-2001
			EP 1052965 A1	22-11-2000
			HU 0100868 A2	30-07-2001
			ID 26224 A	07-12-2000
			JP 2002501877 T	22-01-2002
			US 6177390 B1	23-01-2001
			ZA 9900778 A	03-08-1999
US 5543074	A	06-08-1996	AU 1809795 A	04-09-1995
			BR 9506797 A	16-09-1997
			CA 2180942 A1	24-08-1995
			DE 69519372 D1	14-12-2000
			DE 69519372 T2	26-04-2001
			WO 9522311 A1	24-08-1995
			EP 0744935 A1	04-12-1996
			ES 2153026 T3	16-02-2001
			JP 2746758 B2	06-05-1998
			JP 9501951 T	25-02-1997
US 6090773	A	18-07-2000	ID 18376 A	02-04-1998
			US 2003158065 A1	21-08-2003
			US 6489286 B1	03-12-2002
			AU 718594 B2	20-04-2000
			AU 1709997 A	20-08-1997
			CA 2245010 A1	31-07-1997
			CN 1214628 A	21-04-1999
			DE 69714016 D1	22-08-2002
			DE 69714016 T2	13-03-2003
			EP 0879047 A1	25-11-1998

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 03/12066

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 6090773	A	WO 9726860 A1	31-07-1997